

Name KEY

Date: 5/8/13 Period: \_\_\_\_\_ ASSIGN

EOG Review: Water (Hydrosphere)

Word Bank:

Gulf Stream	3	Ice	Pollution	Upwelling	Phytoplankton (2)
Icebergs	97	Divide	Satellites	Rivers	Impermeable rock
Gas	Glaciers	Freshwater	Ocean	Divers	Mangrove forest
Fresh	Liquid	Lakes	Estuaries	Current	Drainage basins
Groundwater	Salt	Buoyancy	Surface zone	Solid	Lake turnover
71	Polarity	Surface	Sonar	Decreases	Overfishing
29	Cohesion	Water table	Deep zone	Wind	Hydrothermal vents
Intertidal zone	Adhesion	Rivers (2)			

1. Water can exist as a solid, liquid, and gas on Earth.
2. Water molecules stick to other water molecules in cohesion.
3. Water easily dissolves substances with ionic bonds. This is because water has polarity, or a negative end and a positive end.
4. Water molecules stick to other types of molecules in adhesion.
5. Surface tension is caused by a strong attraction between surface molecules.
6. The density of fresh water is less than the density of salt water.
7. Buoyancy is the force of water pushing up on an object to make it float.
8. Freshwater makes up 3 % of all the water on Earth. Saltwater makes up 97 % of all water on Earth.
9. 2/3 of all freshwater is found frozen (ice) in glaciers or icebergs.
10. The top portion of an aquifer that is saturated with water is called a(n) water table.
11. About 29 % of the Earth is covered with land and about 71 % of the Earth's surface is covered with water.
12. On Earth, water flows from divides into areas known as drainage basins (also called: <sup>-river basins</sup> watersheds).
13. Water that collects and moves beneath Earth's surface is called groundwater.
14. As water seeps into the ground, it is stopped by a layer of impermeable rock.
15. An area that determines the direction of water flow is called a(n) divide.
16. Water in aquifers is the most used source of freshwater by people.
17. Two additional sources of fresh water are desalination and icebergs.
18. Rivers drop sediments that contain valuable minerals into the ocean.
19. Organisms that live in the intertidal zone must be able to live in and out of the water.
20. Cracks in the ocean's crust that allow heat from the Earth to escape are called hydrothermal vents.
21. Tiny, plantlike organisms that undergo photosynthesis are called Phytoplankton.
22. Overfishing and Pollution are two negative outcomes from commercial fishing.
23. This type of wetland is found in warm tropical regions and is home to many trees: mangrove forest.
24. Fresh water from rivers meets salt water from the ocean in environments called estuaries.
25. The open ocean is divided into two zones: surface (photic) and deep.
26. The Gulf Stream moves warm water towards Great Britain creating a mild climate.
27. Upwelling (deep) currents move nutrients to the surface and mix oxygen within the ocean.
28. A current is a mass of moving water.
29. As you travel further down into the ocean the temperature decreases (gets colder).
30. Plants and wind mixing water and air are the source of dissolved oxygen in the ocean.
31. How might organisms interact to connect terrestrial and aquatic food webs?  
Land organisms eat things that live in the water. Bears eat fish, racoons eat crabs, etc.
32. How might water pollution affect this combined food web?  
Water pollution might kill aquatic organisms, which might be an important food source
33. How might land pollution affect this combined food web? for land organisms, which may die and overpopulated.  
Some fish may lose predators and become disrupted the land food web too.

Word Bank:

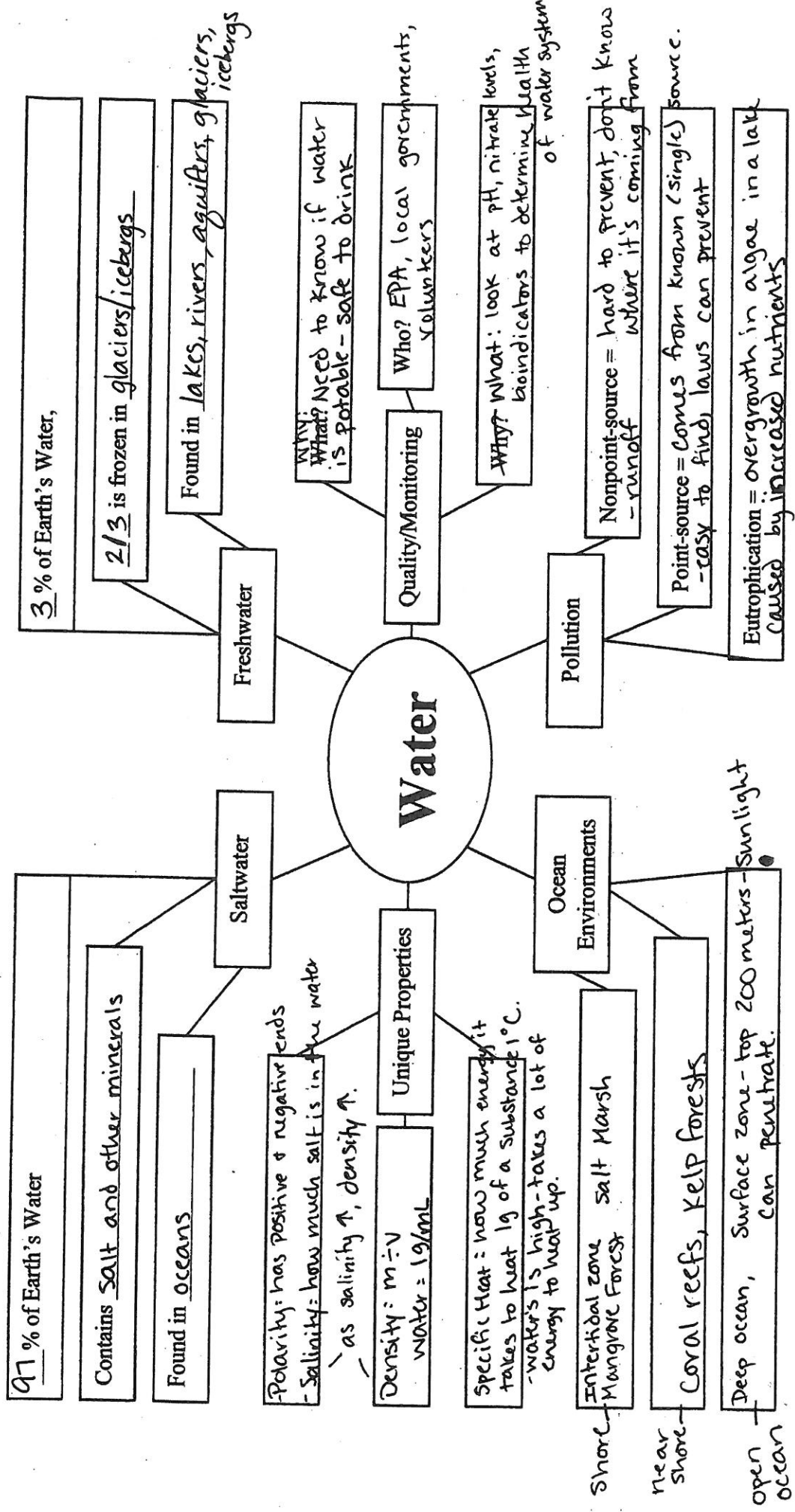
95	Non-point source (2)	Laws	Migrating	EPA	Reuse	Runoff
Water	Eutrophication	Irrigation	Precipitation	Septic systems	Recycle	Hydroelectricity
Runoff	Point-source	Dams	Condensation	Sewage Systems	Desalination	Increases
Easy	Concentrations	Turbidity	Evaporation	Reduce	Chemicals (2)	

34. 95 % of the ocean remains unexplored.
35. Without water there would be no life on Earth.
36. On Earth, water that flows off the side of the land and into a drainage basin is known as runoff.
37. Point-source pollution is easy to locate. Many laws are put into place to prevent this pollution.
38. The type of pollution that causes most water pollution is nonpoint source pollution.
39. An example of Point source pollution is an oil spill.
40. An example of nonpoint source pollution is runoff from nearby farms or yards.
41. Dangerous chemicals are only allowed in low concentrations because of the damage they cause.
42. The use of water to grow crops is called irrigation.
43. Hydroelectricity is electricity created from the moving of water in tides, river, and falls.
44. Turbidity is the measure of how murky or muddy water appears.
45. Dams are built across rivers to help control the water, but interfere with migrating fish.
46. Evaporation, Condensation, and Precipitation are all parts of the water cycle.
47. What is transpiration?  
The evaporation of water from the leaves of plants.
48. The EPA determines whether the levels of chemicals in the water are safe for human consumption.
49. Septic Systems are used to treat wastewater in rural areas, and Sewage Systems treat water in urban areas.
50. We can conserve water in three ways: reduce, reuse, and recycle.
51. The process of removing salt from salt water is known as desalination.
52. Chemicals are added to the water in water treatment plants to kill off any harmful microorganisms.
53. Runoff from land is the source of most pollution in the ocean waters. (nonpoint source)
54. Water needs to be tested to ensure it is NOT contaminated by bacteria or chemicals.
55. As salinity increases, density increases.
56. Describe a river basin.  
Where water collects in low areas-surrounded by high land (divides). Water flows through the river basin to the ocean.
57. How might a bioindicator help us monitor water quality?  
Health of organisms in the water (and biodiversity in the water) shows the health of the water. Lots of healthy organisms shows good water quality.
58. Why are temperature, pH, dissolved oxygen, and nitrate levels important to water quality?  
Must be within a certain range to be safe.
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|----------|----------------------|----|--|
| <u>H</u> | 59. Specific Heat    | a. | How much salt is dissolved in water (salt concentration)   |
| <u>J</u> | 60. Cohesion         | b. | Extreme tides  |
| <u>I</u> | 61. Adhesion         | c. | The science and business of raising fish; fish farming   |
| <u>K</u> | 62. Polarity         | d. | Mass/volume  |
| <u>D</u> | 63. Density          | e. | When molecules have a positive region and negative region  |
| <u>A</u> | 64. Salinity         | f. | Weak tides   |
| <u>F</u> | 65. Neap Tide        | g. | How much solute is dissolved in solvent; measured in ppm   |
| <u>B</u> | 66. Spring Tide      | h. | How much energy it takes to heat a substance; water's is high  |
| <u>G</u> | 67. Concentration    | i. | When water molecules stick to other substances   |
| <u>C</u> | 68. Aquaculture      | j. | When water molecules stick to other water molecules  |
| <u>K</u> | 69. Keystone Species | k. | A species that if you remove it from an environment, all of the other animals in the food web are affected |
70. Explain how water's specific heat affects Earth's climate.  
Water has a high specific heat, meaning it takes longer to change the temperature. Warms atmosphere over ocean, seasons occur later.

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Summarize the information in this diagram.

